



The halogenoalkane 2-bromo-3-methylbutane reacts with hot ethanolic potassium hydroxide to form a mixture of two alkenes and an alcohol.

For the formation of each product, write a balanced equation, state the role of the hydroxide ions, name and draw the mechanism.

Alcohol	<i>balanced equation</i>	
	<i>role of hydroxide ion</i>	nucleophile
	<i>mechanism name</i>	nucleophilic substitution
	<i>mechanism</i>	

Alkene 1	<i>balanced equation</i>	
	<i>role of hydroxide ion</i>	base
	<i>mechanism name</i>	elimination
	<i>mechanism</i>	

Alkene 2	<i>balanced equation</i>	
	<i>role of hydroxide ion</i>	base
	<i>mechanism name</i>	elimination
	<i>mechanism</i>	