Test yourself

Chapter 15

1 What is the name of the following halogenoalkane?

CH₃CH₂CH₂CHBrCH₃

- **A** 4-bromopentane
- **B** 2-bromobutane
- **C** 2-bromopentane
- **D** 1,1-bromomethylbutane
- 2 We can describe CH₃CH₂CH₂CHBrCH₃ as a ...
 - **A** primary halogenoalkane
 - **B** secondary halogenoalkane
 - **C** tertiary halogenoalkane
 - **D** carbocation
- 3 Halogenoalkane X is heated under reflux with aqueous sodium hydroxide solution. The organic product is ethanol. Which one of the following could be the structural formula of X?
 - A CH₃CH₂Cl
 - **B** ClCH₂CH₂Cl
 - C CH₂=CHCl
 - D CH₃Cl

- Which one of the following would undergo hydrolysis at the fastest rate?
 - A CH₃CH₂CH₂CH₂Br
 - B CH₃CH₂CH₂CH₂Cl
 - C CH₃CH₂CH₂CH₂F
 - **D** CH₃CH₂CH₂CH₂I
- What name is given to the reaction in which a halogenoalkane is hydrolysed?
 - **A** Electrophilic addition
 - **B** Electrophilic elimination
 - C Nucleophilic addition
 - **D** Nucleophilic substitution
- Which one of the following statements about the mechanism of hydrolysis of bromoethane and 1-bromopropane is correct?
 - A Both bromoethane and 1-bromopropane are hydrolysed by an S_N 1 mechanism
 - B Both bromoethane and 1-bromopropane are hydrolysed by an S_N 2 mechanism
 - C Bromoethane is hydrolysed by an S_N 2 mechanism and 1-bromopropane is hydrolysed by an S_N 1 mechanism
 - **D** Bromoethane is hydrolysed by an S_N1 mechanism and 1-bromopropane is hydrolysed by an S_N2 mechanism

- 7 1-bromopropane is heated under reflux with ethanolic potassium cyanide. What is the name of the organic product of the reaction?
 - **A** Propylamine
 - **B** Propanenitrile
 - **C** Butanenitrile
 - **D** Butylamine
- **8** What reagents are needed to convert bromoethane to ethene?
 - **A** Potassium dichromate(VI) and dilute sulfuric acid
 - **B** Potassium manganate(VII) with dilute sulfuric acid
 - **C** Ethanolic solution of sodium hydroxide
 - **D** Concentrated ammonia solution
- Which one of the following is **not** a use that has been made of CFCs (chlorofluorocarbons)?
 - **A** Whitener in washing powder
 - **B** Refrigerants
 - **C** Solvents
 - **D** Blowing agents in the manufacture of foam plastics
- 10 CFCs were used as propellants in aerosol cans. Why is their use now banned?
 - A They caused explosions in the presence of a naked flame
 - **B** They cause acid rain
 - **C** They cause depletion of the ozone layer
 - **D** They were found to be toxic to humans